	IBRIDGE INTERNATIONAL EXAMINATIONS
AGRICULTURE	5038/03
Paper 3 Practical Test	
	October/November 2005
Candidates answer on the Que Additional Materials: As listed ir	
Write in dark blue or black pen in the sp You may use a soft pencil for any diagr Do not use staples, paper clips, highligh Answer <b>all</b> questions.	andidate number in the spaces provided at the top of this page. baces provided on the Question Paper. ams, graphs or rough working.
	For Examiner's Us
	1
	2
	3
	Total

Answer **all** the questions.

2

Write your answers in the spaces provided.

- 1 You are going to investigate the effect of different concentrations of sugar solution on strips of potato.
  - Take five strips of potato. Use a scalpel to make each potato strip 50 mm long.
  - Place each strip in the centre of a separate petri dish as shown in Fig. 1.1.

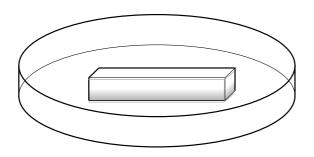


Fig. 1.1

• Label the petri dishes and then fill the dish with the solutions as described below.

petri dish number	1	2	3	4	5
solution	0% sugar solution	10% sugar solution	20% sugar solution	30% sugar solution	AS1

• Leave the petri dishes for 30 minutes.

## You are advised to answer question 2 or 3 while this experiment is working.

- Remove the potato from the petri dish and measure the length of each strip.
- (a) Record your results below.

petri dish	concentration of sugar solution	length of strip/mm
1	0%	
2	10%	
3	20%	
4	30%	
5	AS1	

[5]

(b)	Explain the results you have observed in petri dishes <b>1</b> and <b>4</b> .
	dish 1
	dish <b>4</b>
	[4]
(c)	Why is it important that each potato strip starts the same length?
	[1]
(d)	Suggest, giving a reason for your answer, the concentration of sugar in AS1.
	concentration%
	reason
	[2]
	[Total: 12]

2 Calcium nitrate is an agricultural fertiliser.

You are going to perform a series of tests to identify which of two agricultural chemicals, **AS2** and **AS3** contains calcium nitrate.

ion	tests	test result
ammonium	add sodium hydroxide solution, gently warm, do not allow to boil	ammonia produced
calcium	add sodium hydroxide solution	white precipitate
carbonate	add dilute hydrochloric acid	fizzing, carbon dioxide produced
nitrate	add sodium hydroxide solution then aluminium foil; warm very carefully. Do not allow to boil	ammonia produced
sulphate	add dilute hydrochloric acid then add barium chloride solution	white precipitate

## Table 2.1

(a) Describe how you have performed the tests given in Table 2.1 to identify calcium nitrate.

 5

(b) (i) Describe your results of some of the tests in Table 2.2.

Table	2.2
IGNIC	

-		calcium ion test	nitrate ion test
	AS2		
	AS3		
(ii) F		ults, which agricultural chemical co	[4]
	·		[1]
	um nitrate is v o the farmer.	very soluble in water. Suggest an	advantage and a disadvantage of
advar	ntage		
disad	vantage		[2]

(c)

- 3 You have been provided with two seeds, **AS4**, which have been soaked in water.
  - (a) Carefully remove the outer layer of **one** of the seeds. You may need to use a scalpel to help you, but do not damage the inside.

Make a clear line drawing of the seed showing any structures you can see.

[2]

(b) Using the scalpel, divide the other seed in half to show all of its internal structures.Draw a large, labelled drawing of the cut section showing these internal structures.

[5]

[Total: 7]

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## SUPERVISOR'S REPORT

\* The Supervisor or Teacher responsible for the subject is asked to answer the following questions.

1 Was any difficulty experienced in providing the necessary materials? Give brief details.

- 2 Did the candidate experience any difficulty during the course of the examination? If so, give brief details. Reference should be made to
  - (a) difficulties arising from faulty specimens;
  - (b) accidents to apparatus or materials;
  - (c) any information that is likely to assist the Examiner, especially if this cannot be discovered from the scripts.

3 Name of large bean seed species/variety used for AS4 .....

Declaration to be signed by the Principal, and completed on the top script from the Centre.

The preparation of the Practical Test has been carried out so as to fully maintain the security of the examination.

Signed.....

\*Information that applies to all candidates need only be given once.

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